#### **B.L.D.E.** Association's

# S.B.Arts and K.C.P. Science College Vijayapur

## PG DEPARTMENT OF CHEMISTRY



# Programme Outcomes (POs), Programme Specific Outcomes (PSOs) and Course Outcomes (COs)

#### **B.L.D.E.** Association's

#### S.B. Arts and K.C.P Science College Bijapur

#### Post Graduate Department of Chemistry POS 2019-2020

**Subject: Spectroscopy** 

**PO1:** In advance elementary/fundamental knowledge.

**PO2:** Critical thinking, scientific methods to design, carry out analytical the results of experiments and get awareness of the impact of chemistry on environment, society, etc.

**PO3:** Higher education, competitive, Reputed Research laboratory.

**PO4:** Industrial application.

**PSO1:** To develop strong and compete knowledge in theoretical and practical chemistry.

**PSO2**: Able to explain Theory, Principle, Postulates, Methods, explaining instrumentation, Derivation, calculations and to calculate the physical and electrochemical parameters

**PSO3:** To recognize the various laws and theories and solving numerical problems.

**PSO4:** To develop various technical and analytical skills through laboratory training.

**POS5:** To create awareness the importance. And impact of chemistry on environment.

#### M.Sc 1st Sem: Spectroscopy-I

**CO1:** Review of different types of electromagnetic radiations.

**CO2:** Study the types of transitions and their energy levels.

**CO3:** Understand the selection rules.

**CO4:** Study the classification of polyatomic molecules (CO<sub>2</sub>, CH<sub>3</sub>F and BCl<sub>3</sub>) based on moment of inertia-linear, symmetric top and asymmetric top.

**CO5:** To know the detail study of UV-Visible Spectroscopy.

**CO6:** To study the  $\lambda$ max for polyenes,  $\alpha$ , $\beta$ -unsaturated aldehydes and ketones (Woodward-Fisher rules), aromatic systems and their derivatives.

**CO7:** To know about the number of degrees of freedom of vibration, modes of vibratioa and, Vibrational coupling overtones and Fermi resonance.

**CO8:** To study the brief discussion of identification of functional groups alkanes, alkenes, aromatics, carboxylic acids, carbonyl compounds(aldehydes and ketones, esters), amides and amines.

CO9: To study the principle, instrumentation and applications of Raman Spectra

**COURSE**: M.Sc Ist Semester (Theory)

Course Code : CHGT-1.4 Subject: Spectroscopy-I

Course	P01	P02	P03	P04	PSO1	PSO2	PSO3	PSO4	PSO5
Outcomes									
CO1	2	2	3	3	1	3	3	2	3
CO2	1	1	3	2	3	3	3	2	3
CO3	-	2	1	3	3	3	3	2	3
CO4	-	3	2	2	3	3	3	2	3
CO5	3	2	1	3	3	3	3	2	3
C06	2	3	2	1	3	3	3	2	3
CO7	2	1	1	3	3	3	3	2	3
C08	-	3	2	1	3	3	3	2	3
CO9	-	3	1	1	3	3	3	2	3

#### M.Sc 2<sup>nd</sup> Sem: Spectroscopy-II

**CO1:** To understand the magnetic properties of nuclei.

**CO2:** To learn about the various factors influencing in NMR spectroscopy.

**CO3:** To know about the principle, instrumentation and applications of FT-NMR spectroscopy.

**CO4:** To study the brief discussion of simplification of complex spectra.

**CO5:** To know the detail study of the <sup>13</sup>C-NMR spectroscopy.

**CO6:** To learn about the two dimensional NMR spectroscopy (COSY, NOESY, DEPT Spectra and MRI).

**CO7:** To know the detail study of the mass spectroscopy.

**CO8:** To understand the basic theory, principle and instrumentation of different mass spectroscopy techniques.

**CO9:** To know about the modes of fragmentation and their rules for different class of organic compounds.

**CO10:** Combined applications of spectroscopic techniques.

**COURSE**: M.Sc II<sup>nd</sup> Semester (Theory)

Course Code : CHGT-2.4 Subject: Spectroscopy-II

Course	P01	P02	P03	P04	PSO1	PSO2	PSO3	PSO4	PSO5
Outcomes									
CO1	3	2	3	1	3	3	3	2	3
CO2	1	3	2	2	3	3	3	2	3
CO3	3	2	3	3	3	3	3	2	3
CO4	1	3	-	2	3	3	3	2	3
CO5	2	3	2	1	3	3	3	2	3
C06	1	3	2	2	3	3	3	2	3
CO7	-	2	1	3	3	3	3	2	3
C08	2	3	2	3	3	3	3	2	3
CO9	2	3	2	2	3	3	3	2	3
CO10	1	3	1	3	3	3	3	2	3

#### M.Sc 3rd Sem: Spectroscopy-III

**CO1:** To study the basic applications of infra red spectroscopy to inorganic compounds.

**CO2:** To know the changes in infrared spectra of donor molecules upon coordination.

**CO3:** To learn about the change in spectra accompanying change in symmetry upon coordination.

**CO4:** To know the detail study of the FTIR.

**CO5:** To learn about basic principle and interaction between spin and magnetic field ESR spectroscopy.

**CO6:** To discuss the various factors affecting for ESR spectroscopy.

**CO7:** To know the detail study of the nuclear quadrupole resonance spectroscopy.

**CO8:** To study the theory, principles and experimental methods of mossbauer spectroscopy.

**COURSE**: M.Sc III<sup>rd</sup> Semester (Theory)

Course Code : CHGT-3.4 Subject: Spectroscopy-III

Course	P01	P02	P03	P04	PSO1	PSO2	PSO3	PSO4	PSO5
Outcomes									
CO1	1	3	2	1	3	3	3	2	3
CO2	2	3	3	3	3	3	3	2	3
CO3	1	2	2	2	3	3	3	2	3
CO4	-	2	2	3	3	3	3	2	3
CO5	3	3	3	1	3	3	3	2	3
C06	3	1	2	2	3	3	3	2	3
CO7	2	2	3	3	3	3	3	2	3
C08	-	1	2	2	3	3	3	2	3

#### M.Sc IVth Sem: Spectroscopy-IV

**CO1:** To know the detail study of the flame emission spectroscopy.

**CO2:** To understand the basic principle, theory and flame spectra variation of emission intensity with flames, flame background, metallic spectra in flame.

**CO3:** To study the applications of flame emission spectroscopy.

**CO4:** To know the detail study of the chiroptical spectroscopy.

**CO5:** To learn about the plane polarized light, instrumentation and optical rotary dispersion (ORD) of chiroptical spectroscopy.

**CO6:** To determine the configuration of cyclic and steroidal ketones.

**CO7:** To study the theoretical basics for fluorescence and phosphorescence in molecular luminescence spectroscopy.

**CO8:** General scope of applications of luminescence.

**CO9:** To know the detail study of the photoelectron spectroscopy.

**CO10:** To learn about the X-ray photoelectron, Auger electron spectroscopy and applications.

**COURSE**: M.Sc IV<sup>th</sup> Semester (Theory)

Course Code : CHGT-4.4 Subject: Spectroscopy-IV

Course	P01	P02	P03	P04	PSO1	PSO2	PSO3	PSO4	PSO5
Outcomes									
CO1	1	-	3	3	3	3	3	2	3
CO2	-	1	2	3	3	3	3	2	3
CO3	2	3	1	3	3	3	3	2	3
CO4	1	3	2	3	3	3	3	2	3
CO5	2	2	2	1	3	3	3	2	3
C06	2	1	1	2	3	3	3	2	3
CO7	1	2	3	3	3	3	3	2	3
C08	2	1	2	3	3	3	3	2	3
CO9	2	3	1	3	3	3	3	2	3
CO10	1	3	2	3	3	3	3	2	3

P. G. Department of Chamistry, SB Arts & KCP Science Ladlege BIJAPUR - 586101

IQAe, Co-ordinator S.B.Arts & K.C.P.Science College, Vijayapur. Principal, S.B.Arts & K.C.P. Science College BIJAPUR.

### **EVALUATION MAPPING**

#### **THEORY:**

Marks Distribution:

- 1. Internal Assessment = 10 marks
- 2. University Examination = 40 marks

Sl No	Parameter	Percentage (%)
1	Knowledge	10
2	Understanding	15
3	Numericals	05
4	Descriptive	20

P. G. Denartment of Chamistry, SB Arts & KCP Science Lallege BUAFUR - 586101

IQAC, Co-ordinator
S.B.Arts & K.C.P.Science College,
Vijayapur.

Principal,
S.B.Arts & K.C.P. Science College
BIJAPUR.